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# **Fluid Mechanics**

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• Pressure is the (compression) force exerted by a fluid per unit area.

$$Pressure = \frac{Force}{Area} \quad \left(\frac{N}{m^2}\right) \equiv Pa$$

- Stress vs. pressure?
- In fluids, gases and liquids, we speak of pressure; in solids this is normal stress. For a fluid at rest, the pressure at a given point is the same in all directions.



 Differences or gradients in pressure drive a fluid flow, especially in ducts and pipes.

The density of a fluid is its mass per unit volume:

$$\rho = \frac{m}{v} \left( \frac{kg}{m^2} \right)$$

- Liquids are essentially incompressible
- Density is highly variable in gases nearly proportional to the pressure.

@20°C, 1 atm	Air	Water	Hydrogen	Mercury
Density (kg/m <sup>3</sup> )	1.20	998	0.0838	13,580

• Note: *specific volume* is defined as:

$$v = \frac{V(m^3)}{m(kg)} = \frac{1}{\rho}$$

# Specific weight

• The specific weight of a fluid is its weight, , per unit volume. Density and specific weight are related by gravity:

#### Specific gravity

Specific gravity is the ratio of a fluid density to a standard reference fluid, typically water at 4°C (for liquids) and air (for gases):

 $\gamma = \rho g$ 

$$SG_{gas} = \frac{\rho_{gas}}{\rho_{air}} = \frac{\rho_{gas}}{1.205 \ (kg/m^3)}$$
$$SG_{liquid} = \frac{\rho_{liquid}}{\rho_{water}} = \frac{\rho_{liquid}}{1000 \ (kg/m^3)}$$

• For example, the specific gravity of mercury is  $SG_{Hg}$ = 13,580/1000  $\degree$ 13.6.

# Kinetic and potential energy

• <u>Potential energy</u> is the work required to move the system of mass *m* from the origin to a position against a gravity field *g*:



PE = mgz

<u>Kinetic energy</u> is the work required to change the speed of the mass from zero to velocity *V*.

$$KE = \frac{1}{2}mV^2$$

<u>Note</u>: internal energy, *u*, is a function of temperature and pressure for the single-phase substance, whereas KE and PE are kinematic quantitie

# Specific heat

- <u>Specific heat capacity</u>: is the measure of the heat energy required to increase the temperature of a unit mass of a substance by one degree temperature.
- *c*<sub>aluminum</sub>=0.9 (kJ/kg.K) and *c*<sub>water</sub>= 4.186 ( kJ/kg.K )
- There are two types of specific heats, constant volume c<sub>ν</sub>and constant pressure c<sub>ρ</sub>.



- Any equation that relates the pressure, temperature, and specific volume of a substance is called an equation of state.
- It is experimentally observed that at a low pressure the volume of a gas is

proportional to its temperature:



 $R_u$  is the gas universal constant,  $R_u$  = 8.314 (kJ/kmol.K)

The constant *R* is different for each gas; for air, R<sub>air</sub>= 0.287 (*kJ/kg.K*). The molecular weight of air *M*=28.97 *kg/kmol*.

#### Properties of ideal gas

• For an ideal gas, *internal energy* is only a function of temperature; thus <u>constant volume specific heat</u> is only a function of temperature:

$$c_{v} = \left(\frac{\partial u}{\partial T}\right)\Big|_{v} = \frac{du}{dT} = c_{v}(T)$$
$$du = c_{v}(T)dT$$

 For an ideal gas, enthalpy is only a function of temperature; h = u + pv• The constant pressure specific heat can be defined as:

$$c_{p} = \left(\frac{\partial h}{\partial T}\right)\Big|_{p} = \frac{dh}{dT} = c_{p}(T)$$
$$dh = c_{p}(T)dT$$
$$R = c_{p} - c_{p}$$

The specific heat ratio is an important dimensionless parameter:

$$k = \frac{c_p}{c_v} = k(T) \ge 1$$

#### Incompressible fluids

• Liquids are (almost) incompressible and thus have a single constant specific heat:

$$c_v = c_v = c$$
  $dh = cdT$ 

# Viscosity

• Viscosity is a measure of a fluid's resistance to flow. It determines the fluid strain rate that is generated by a given applied shear stress.



• Temperature has a strong and pressure has a moderate effect on viscosity. The viscosity of gases and most liquids increases slowly with pressure.

$$\begin{split} \mu_{hydrogen} &= 9.0E - 6 \left(\frac{kg}{m.s}\right) \qquad \mu_{air} = 1.8E - 5 \left(\frac{kg}{m.s}\right) \\ \mu_{water} &= 1.0E - 3 \left(\frac{kg}{m.s}\right) \qquad \mu_{engine\ o\ il,SAE30} = 0.20 \left(\frac{kg}{m.s}\right) \end{split}$$

# Viscosity

• A <u>Newtonian fluid</u> has a linear relationship between shear stress and



- The *no-slip condition*: at the wall velocity is zero relative to the wall. This is a characteristic of all viscous fluid.
- The shear stress is proportional to the slope of the velocity profile and is greatest at the wall.

## The Reynolds number

• The Reynolds number, Re, is a dimensionless number that gives a measure of

$$Re = \frac{\rho V L}{\mu} = \frac{V L}{\nu}$$



the ratio of inertial forces

to viscous forces



Creeping flow, Re is very low

Laminar flow, Re moderate Turbulent flow, Re high

#### Thermal conductivity

• Rate of heat conduction is proportional to the temperature difference, but it is inversely proportional to the thickness of the layer



- To make this equality, k (W/m.K)the thermal conductivity of the material, is introduced.
- This is called the Fourier's law of heat conduction:

$$\begin{aligned} q &= \frac{Q}{A} = -k\nabla T \\ q_x &= -k\frac{\partial T}{\partial x}, \quad q_y = -k\frac{\partial T}{\partial y}, \quad q_z = -k\frac{\partial T}{\partial z} \end{aligned}$$

#### Flow between parallel plates

• It is the flow induced between a fixed lower plate and upper plate moving steadily at velocity *V* 



• Shear stress is constant throughout the fluid:

$$\frac{du}{dy} = \frac{\tau}{\mu} = const.$$

• After integration and applying boundary conditions:



• A liquid, being unable to expand freely, will form an *interface* with a second liquid or gas.



• The cohesive forces between liquid molecules are responsible for the phenomenon known as surface tension.

- Surface tension Y (pronounced upsilon) has the dimension of force per unit length (N/m) or of energy per unit area  $(J/m^2)$ .
- Yair-water = 0.073 N/m; Yair-mercury = 0.48 N/m

# (a)

#### Surface tension

 Using a force balance, pressure increase in the interior of a liquid half cylinder droplet of length L and radius R is:

$$2RL\Delta p = 2\Upsilon L \quad or \quad \Delta p = \frac{\Upsilon}{R}$$

 Contact angle θ appears when a liquid interface intersects with a solid surface.

$$\theta = \begin{cases} < 90^{\circ} & wetting liquid \\ > 90^{\circ} & nonwetting liquid \end{cases}$$

 Water is extremely wetting to a clean glass surface with θ≈0. For a clean mercury-air-glass interface, θ≈130°.

#### Vapor pressure and cavitation

- Vapor pressure: the pressure at which a liquid boils and is in equilibrium with its own vapor.
- When the liquid pressure is dropped below the vapor pressure due to a flow phenomenon, we call the process *cavitation*.



• The dimensionless parameter describing flow-induced boiling is called cavitation number:

$$Ca = \frac{p_a - p_v}{0.5\rho V^2}$$





Bubble formation due to high velocity (flow-induced boiling).



Damage (erosion) due to cavitation on a marine propeller.

### No-slip and no-temp jump

• When a fluid flow is bounded by a surface, molecular interactions cause the fluid in contact with the surface to seek momentum and energy equilibrium

with the surface.



$$V_{fluid} = V_{wall}$$
 No-slip flow condition  
 $T_{fluid} = T_{wall}$  No-temperature jump condition  
Speed of sound & compressibility

• the compressibility effects are important at high gas flows due to significant



density changes.

• Speed of sound: is the rate of propagation of small disturbance pressure pulses (sound waves) through the fluid:

$$a^{2} = k \left(\frac{\partial p}{\partial \rho}\right)_{T}, \qquad k = \frac{c_{p}}{c_{v}}$$

- For an ideal gas Atdraiger = \langle kRT
- Mach number is the ratio of the flow to the speed of sound

 $Ma = \frac{V}{a}$ 

• Compressibility effects are normally neglected for Ma<0.3



- <u>Streamline</u>: is a line everywhere tangent to the velocity vector at a given instant.
- <u>Pathline</u>: is the actual path traversed by a given fluid particle.
- **Note**: in steady flows, streamlines and pathlines are identical.

If the elemental arc length *dr*of a streamline is to be parallel to *V*, their respective components must be in proportion:

 $\frac{dx}{u} = \frac{dy}{v} = \frac{dz}{w} = \frac{dr}{V}$ 

Bernoulli equation takes the form of

$$\frac{V_1^2}{2} + \frac{P_1}{\rho} + gz_1 = \frac{V_2^2}{2} + \frac{P_2}{\rho} + gz_2$$

where V is the fluid velocity, P is the fluid pressure, z is the elevation of the location in the pipe relative to a specified reference elevation (datum),  $\rho$  is the fluid density, and g is gravity



The velocities at two axial locations in the duct with different areas are related through the conservation of mass equation,

$$\rho_1 V_1 A_1 = \rho_2 V_2 A_2 = \dot{m}$$

 $P_{1}$   $A_{1} z_{1}$   $V_{1}$   $V_{1}$   $V_{1}$   $V_{1}$   $P_{2}$   $A_{2} z_{2}$   $A_{2} z_$ 

is usually written in the form:  $V_1A_1 = V_2A_2 = Q$  where Q is the volume flow rate (e.g., m<sup>3</sup>/s).

Equations can be combined to obtain an expression

$$V_2 = \frac{1}{\sqrt{1 - (A_2/A_1)^2}} \sqrt{\frac{2[(P_1 + g\rho z_1) - (P_2 + g\rho z_2)]}{\rho}}$$

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$$V_1A_1 = V_2A_2 = Q$$

$$Q = \frac{A_2}{\sqrt{1 - (A_2/A_1)^2}} \sqrt{\frac{2[(P_1 + g\rho z_1) - (P_2 + g\rho z_2)]}{\rho}}$$

The <u>theoretical</u> basis for a class of flow meters in which the flow rate is determined from the pressure change caused by variation in the area of a conduit.





$$Q = \frac{CA_2}{\sqrt{1 - (A_2/A_1)^2}} \sqrt{\frac{2[(P_1 + g\rho z_1) - (P_2 + g\rho z_2)]}{\rho}}$$

The factor C, called the discharge coefficient

is used to account for nonideal effects.

and a parameter called the Reynolds number, which is defined as

$$\operatorname{Re} = \frac{\rho V D}{\mu}$$

$$Q = \frac{CA_2}{\sqrt{1 - (A_2/A_1)^2}} \sqrt{\frac{2[(P_1 + g\rho z_1) - (P_2 + g\rho z_2)]}{\rho}}$$

$$K = \frac{C}{\sqrt{1 - (A_2/A_1)^2}}$$

K, called the flow coefficient,

When  $Z_1 = Z_2$  Flow Rate Equation becomes as follows:

$$Q = \frac{CA_2}{\sqrt{1 - (A_2/A_1)^2}} \sqrt{\frac{2\Delta P}{\rho}}$$